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Dichlorido[(*Z*)-4-(2,6-diisopropylanilino)pent-3-en-2-one]dimethyltin(IV)

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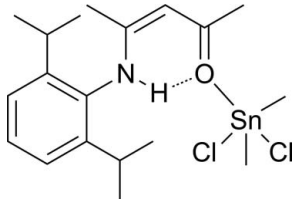
Key indicators: single-crystal X-ray study; T = 297 K; mean σ(C–C) = 0.007 Å; R factor = 0.046; wR factor = 0.104; data-to-parameter ratio = 17.7.

In the crystal structure of the title compound, $[Sn(CH_3)_2Cl_2(C_{17}H_{25}NO)]$, the Sn atom adopts a trigonal-bipyramidal geometry with the O and one Cl atom in axial positions. A weak intramolecular N—H···O hydrogen bond occurs. The crystal structure displays weak intermolecular C—H···Cl interactions.

Related literature

For dichloridodiorganotin(IV) complexes, see: Cunningham *et al.* (2004); Curnow *et al.* (2006); Ianelli *et al.* (1993); Mahadevan *et al.* (1982); Ng (1996); Papadaki *et al.* (2008); Tian *et al.* (2006); Valle *et al.* (1982).



Experimental

Crystal data

$[Sn(CH_3)_2Cl_2(C_{17}H_{25}NO)]$
M_r = 479.04
Triclinic, $P\overline{1}$
a = 8.504 (4) Å
b = 10.212 (4) Å
c = 14.507 (6) Å
α = 71.070 (7)°
β = 83.300 (8)°
γ = 76.984 (7)°
V = 1159.8 (9) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 1.34 mm^{-1}
T = 297 K
0.36 × 0.35 × 0.33 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2000)
T_{min} = 0.626, T_{max} = 0.645
8415 measured reflections
4044 independent reflections
3433 reflections with $I > 2\sigma(I)$
R_{int} = 0.028

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.046
$wR(F^2)$ = 0.104
S = 1.06
4044 reflections
229 parameters
1 restraint
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.57 e Å^{-3}
$\Delta\rho_{min}$ = −0.41 e Å^{-3}

Table 1
Selected geometric parameters (Å, °).

C18—Sn1	2.095 (5)	Cl2—Sn1	2.4644 (17)
C19—Sn1	2.105 (5)	O1—Sn1	2.375 (3)
Cl1—Sn1	2.3478 (16)		
C18—Sn1—C19	142.9 (3)	C19—Sn1—Cl2	94.78 (18)
Cl1—Sn1—O1	81.74 (9)	Cl1—Sn1—Cl2	95.82 (6)
C18—Sn1—Cl2	94.04 (15)	O1—Sn1—Cl2	176.81 (8)

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···O1	0.83 (3)	2.03 (4)	2.662 (5)	133 (4)
C8^{i}—H8^{i}···Cl1	0.93	2.91	3.695 (4)	143
C17^{ii}—H17B^{ii}···Cl2	0.96	2.89	3.783 (6)	155
C19^{iii}—H19C^{iii}···Cl2	0.96	2.94	3.700 (6)	137

Symmetry codes: (i) $-x+1, -y+1, -z+2$; (ii) $x+1, y-1, z$; (iii) $-x+2, -y+1, -z+1$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT-Plus* (Bruker, 2001); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg, 2009); software used to prepare material for publication: *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NC2173).

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supplementary materials

Acta Cryst. (2010). E**66**, m130 [**doi:10.1107/S1600536809055810**]

Dichlorido[(*Z*)-4-(2,6-diisopropylanilino)pent-3-en-2-one]dimethyltin(IV)

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Comment

In our attempts to prepare compounds of type $Me_2R'_2Sn$, where R' = ketinimate ligand, starting from RLi and Me_2SnCl_2, accidental hydrolysis of the lithium derivative lead to the formation of the title complex. A rational preparation of the complex was acomplished later starting from R'H and Me_2SnCl_2.

In the structure of the title compound the geometry around the tin can be described as distorted trigonal bipyramidal, with the Cl(2) and O(1) atoms occupying the axial positions (Fig. 1). The equatorial plane is formed by the two methyl carbon atoms C(18) and C(19) and the Cl(1) atom.

A weak intramolecular hydrogen bond exist between the hydrogen atom bonded to nitrogen and the oxygen atom (Table 2). There are additional weak C–H···Cl interactions (Table 2).

Experimental

A solution of Me_2SnCl_2 in Et_2O (0.84 g, 3.82 mmol) was added to a stirred solution of (*Z*)-4-[(2,6-diisopropylphenyl)amino]pent-3-en-2-one (1 g, 3.85 mmol) in 50 ml Et_2O resulting in a clear red-brown solution. The reaction mixture was stirred for 24 h and than the solvent was removed under reduced presure to give the title compound as a white-yellow powder. Crystals were obtained by slow diffusion of hexane into a dichloromethane solution of the title compound. Yield: 0.6 g (33%). mp = 124–125 °C.

^{1}H NMR ($CDCl_3$, 300 MHz): δ 1.15 [d, 6H$_A$, –CH(C*H*$_3$)$_2$, ^{3}J(H,H) = 6.8 Hz], 1.19 [s, 6H, SnC*H*$_3$, ^{2}J(^{117}Sn,H) = 75.3, ^{2}J(^{119}Sn,H) = 78.7 Hz], 1.20 [d, 6H$_B$, –CH(C*H*$_3$)$_2$, ^{3}J(H,H) = 6.9 Hz], 1.67 [s, 3H, C*H*$_3$C(N)], 2.11 [s, 3H, C*H*$_3$C(O)], 2.92 [sept, 2H, –C*H*(CH$_3$)$_2$, ^{3}J(H,H) = 6.9 Hz], 5.22 [s, 1H, –C*H*-], 7.18 [d, 2H, *H*$_{8,10}$, ^{3}J(H,H) = 7.7 Hz], 7.32 [t, 1H, *H*$_9$, ^{3}J(H,H) = 7.7 Hz], 11.83 [s, 1H, –N*H*-].

^{13}C NMR ($CDCl_3$, 75.5 MHz): δ 10.73 [s, Sn*C*H$_3$, ^{1}J(^{117}Sn,*C*) = 587.1, ^{1}J(^{119}Sn,*C*) = 614.4 Hz], 19.49 [s, *C*H$_3$C(N)], 22.57 [s, –CH(*C*H$_3$)$_2$, (B)], 24.45 [s, –CH(*C*H$_3$)$_2$, (A)], 28.01 [s, -*C*H(CH$_3$)$_2$], 28.43 [s, *C*H$_3$C(O)], 96.31 [s, -*C*H–], 123.68 [s, *C*$_{8,10}$], 128.81 [s, *C*$_9$], 132.27 [s, *C*$_6$], 145.53 [s, *C*$_{7,11}$], 166.74 [s, CH$_3$*C*(N)], 193.59 [s, CH$_3$*C*(O)].

^{119}Sn NMR ($CDCl_3$, 111.9 MHz): δ -1.33.

Refinement

The C-H H atoms were placed in calculated positions (methyl H atoms allowed to rotate but not to tip) with U_{iso}(H) = 1.2U_{eq}(C) (1.5 for methyl H atoms). The N-H H atom was located in a difference map and its position was refined with istotropic displacement parameters with a restrained N–H distance of 0.86 Å.

Figures

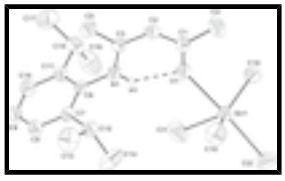


Fig. 1. Crystal structure of the title compound with labelling and displacement ellipsoids drawn at 30% probability level. Hydrogen atoms, except that bonded to nitrogen, were omitted for clarity. Intramolecular hydrogen bonding is shown as a dashed line.

Dichlorido[(*Z*)-4-(2,6-diisopropylanilino)pent-3-en-2-one]dimethyltin(IV)

Crystal data

$[Sn(CH_3)_2Cl_2(C_{17}H_{25}NO)]$	$Z = 2$
$M_r = 479.04$	$F(000) = 488$
Triclinic, $P\overline{1}$	$D_x = 1.372$ Mg m^{-3}
Hall symbol: -P 1	Melting point = 397–398 K
$a = 8.504$ (4) Å	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$b = 10.212$ (4) Å	Cell parameters from 2254 reflections
$c = 14.507$ (6) Å	$\theta = 2.2$–$20.4°$
$\alpha = 71.070$ (7)°	$\mu = 1.34$ mm^{-1}
$\beta = 83.300$ (8)°	$T = 297$ K
$\gamma = 76.984$ (7)°	Block, colourless
$V = 1159.8$ (9) Å^3	0.36 × 0.35 × 0.33 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer	4044 independent reflections
Radiation source: fine-focus sealed tube	3433 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.028$
φ and ω scans	$\theta_{max} = 25°$, $\theta_{min} = 2.2°$
Absorption correction: multi-scan (*SADABS*; Bruker, 2000)	$h = -10 \rightarrow 10$
$T_{min} = 0.626$, $T_{max} = 0.645$	$k = -12 \rightarrow 12$
8415 measured reflections	$l = -17 \rightarrow 17$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods

Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.046$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.104$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.06$	$w = 1/[\sigma^2(F_o^2) + (0.045P)^2 + 0.2662P]$ where $P = (F_o^2 + 2F_c^2)/3$
4044 reflections	$(\Delta/\sigma)_{max} = 0.001$
229 parameters	$\Delta\rho_{max} = 0.57$ e Å^{-3}
1 restraint	$\Delta\rho_{min} = -0.41$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
C1	0.6984 (5)	0.9135 (5)	0.5930 (3)	0.0605 (11)
C2	0.6594 (5)	1.0064 (4)	0.6472 (3)	0.0541 (10)
H2	0.6908	1.0931	0.6209	0.065*
C3	0.5787 (5)	0.9812 (4)	0.7360 (3)	0.0494 (10)
C4	0.7852 (8)	0.9572 (6)	0.4950 (4)	0.101 (2)
H4A	0.8739	0.8825	0.4899	0.152*
H4B	0.8252	1.0408	0.4879	0.152*
H4C	0.7119	0.9762	0.4446	0.152*
C5	0.5455 (7)	1.0906 (5)	0.7874 (4)	0.0781 (15)
H5A	0.4309	1.1209	0.7959	0.117*
H5B	0.592	1.17	0.7494	0.117*
H5C	0.5922	1.0511	0.8501	0.117*
C6	0.4417 (5)	0.8299 (4)	0.8734 (3)	0.0486 (10)
C7	0.5269 (5)	0.7605 (4)	0.9578 (3)	0.0527 (10)
C8	0.4401 (6)	0.7330 (5)	1.0457 (3)	0.0603 (11)
H8	0.494	0.685	1.103	0.072*
C9	0.2759 (6)	0.7751 (5)	1.0501 (3)	0.0641 (12)
H9	0.2195	0.7569	1.1103	0.077*
C10	0.1943 (5)	0.8433 (5)	0.9674 (3)	0.0648 (12)
H10	0.0826	0.872	0.9716	0.078*
C11	0.2755 (5)	0.8706 (5)	0.8765 (3)	0.0565 (11)
C12	0.7095 (5)	0.7131 (5)	0.9532 (4)	0.0650 (12)

H12	0.7507	0.774	0.8923	0.078*
C13	0.7902 (7)	0.7273 (8)	1.0355 (5)	0.109 (2)
H13A	0.7611	0.6615	1.0958	0.163*
H13B	0.7556	0.8218	1.0392	0.163*
H13C	0.9053	0.7078	1.0239	0.163*
C14	0.7528 (8)	0.5650 (7)	0.9482 (6)	0.120 (2)
H14A	0.7234	0.5013	1.0095	0.18*
H14B	0.867	0.5408	0.9348	0.18*
H14C	0.6957	0.558	0.8972	0.18*
C15	0.1811 (6)	0.9386 (6)	0.7843 (4)	0.0727 (14)
H15	0.2549	0.979	0.7309	0.087*
C16	0.1201 (8)	0.8259 (8)	0.7587 (4)	0.112 (2)
H16A	0.2096	0.752	0.7532	0.169*
H16B	0.068	0.8675	0.6977	0.169*
H16C	0.0444	0.7874	0.809	0.169*
C17	0.0427 (8)	1.0556 (8)	0.7938 (5)	0.132 (3)
H17A	−0.0347	1.0172	0.843	0.198*
H17B	−0.0081	1.1	0.7326	0.198*
H17C	0.0825	1.1242	0.8119	0.198*
C18	0.6696 (7)	0.6652 (5)	0.4531 (3)	0.0799 (15)
H18A	0.5909	0.7496	0.4515	0.12*
H18B	0.6202	0.6001	0.4372	0.12*
H18C	0.7568	0.6886	0.4065	0.12*
C19	0.9525 (7)	0.5569 (7)	0.6771 (4)	0.1025 (19)
H19A	0.9978	0.4592	0.7077	0.154*
H19B	0.9138	0.6023	0.7262	0.154*
H19C	1.0339	0.6022	0.6356	0.154*
Cl1	0.5420 (2)	0.51095 (17)	0.70215 (11)	0.1040 (5)
Cl2	0.8568 (2)	0.33755 (15)	0.57138 (13)	0.1102 (6)
H1	0.547 (5)	0.801 (3)	0.752 (3)	0.055 (13)*
N1	0.5289 (4)	0.8616 (4)	0.7806 (3)	0.0513 (8)
O1	0.6627 (4)	0.7925 (3)	0.6232 (2)	0.0761 (10)
Sn1	0.75936 (4)	0.57249 (3)	0.59303 (2)	0.06287 (15)

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.064 (3)	0.058 (3)	0.056 (3)	−0.011 (2)	0.010 (2)	−0.019 (2)
C2	0.060 (3)	0.047 (2)	0.056 (3)	−0.018 (2)	0.003 (2)	−0.014 (2)
C3	0.048 (2)	0.046 (2)	0.057 (3)	−0.0075 (19)	−0.0061 (19)	−0.019 (2)
C4	0.131 (5)	0.096 (4)	0.077 (4)	−0.048 (4)	0.042 (3)	−0.027 (3)
C5	0.111 (4)	0.056 (3)	0.078 (3)	−0.024 (3)	0.013 (3)	−0.036 (3)
C6	0.052 (2)	0.046 (2)	0.053 (2)	−0.0107 (19)	0.004 (2)	−0.024 (2)
C7	0.056 (3)	0.050 (2)	0.057 (3)	−0.011 (2)	−0.002 (2)	−0.023 (2)
C8	0.072 (3)	0.059 (3)	0.050 (3)	−0.015 (2)	−0.001 (2)	−0.016 (2)
C9	0.073 (3)	0.069 (3)	0.052 (3)	−0.019 (3)	0.017 (2)	−0.024 (2)
C10	0.052 (3)	0.074 (3)	0.067 (3)	−0.014 (2)	0.004 (2)	−0.023 (3)
C11	0.060 (3)	0.059 (3)	0.054 (3)	−0.014 (2)	0.002 (2)	−0.022 (2)

C12	0.055 (3)	0.065 (3)	0.074 (3)	−0.008 (2)	−0.004 (2)	−0.022 (2)
C13	0.069 (4)	0.152 (6)	0.125 (5)	−0.024 (4)	−0.020 (4)	−0.062 (5)
C14	0.087 (4)	0.093 (5)	0.188 (7)	0.020 (4)	−0.030 (4)	−0.070 (5)
C15	0.055 (3)	0.094 (4)	0.064 (3)	−0.012 (3)	−0.008 (2)	−0.019 (3)
C16	0.117 (5)	0.156 (7)	0.081 (4)	−0.052 (5)	−0.023 (4)	−0.035 (4)
C17	0.103 (5)	0.141 (7)	0.118 (6)	0.040 (5)	−0.030 (4)	−0.028 (5)
C18	0.100 (4)	0.074 (3)	0.058 (3)	0.002 (3)	−0.021 (3)	−0.017 (3)
C19	0.097 (4)	0.121 (5)	0.101 (5)	−0.013 (4)	−0.029 (4)	−0.048 (4)
Cl1	0.1360 (13)	0.1014 (11)	0.0782 (9)	−0.0581 (10)	0.0188 (9)	−0.0174 (8)
Cl2	0.1549 (15)	0.0523 (8)	0.1206 (13)	−0.0019 (9)	−0.0277 (11)	−0.0282 (8)
N1	0.058 (2)	0.046 (2)	0.055 (2)	−0.0101 (17)	0.0052 (17)	−0.0250 (18)
O1	0.106 (3)	0.061 (2)	0.068 (2)	−0.0291 (19)	0.0276 (19)	−0.0314 (17)
Sn1	0.0865 (3)	0.0497 (2)	0.0515 (2)	−0.01062 (16)	−0.00813 (16)	−0.01462 (15)

Geometric parameters (Å, °)

C1—O1	1.264 (5)	C13—H13A	0.96
C1—C2	1.382 (6)	C13—H13B	0.96
C1—C4	1.502 (6)	C13—H13C	0.96
C2—C3	1.362 (5)	C14—H14A	0.96
C2—H2	0.93	C14—H14B	0.96
C3—N1	1.322 (5)	C14—H14C	0.96
C3—C5	1.494 (6)	C15—C17	1.506 (8)
C4—H4A	0.96	C15—C16	1.524 (8)
C4—H4B	0.96	C15—H15	0.98
C4—H4C	0.96	C16—H16A	0.96
C5—H5A	0.96	C16—H16B	0.96
C5—H5B	0.96	C16—H16C	0.96
C5—H5C	0.96	C17—H17A	0.96
C6—C11	1.381 (6)	C17—H17B	0.96
C6—C7	1.392 (6)	C17—H17C	0.96
C6—N1	1.435 (5)	C18—Sn1	2.095 (5)
C7—C8	1.376 (6)	C18—H18A	0.96
C7—C12	1.519 (6)	C18—H18B	0.96
C8—C9	1.366 (6)	C18—H18C	0.96
C8—H8	0.93	C19—Sn1	2.105 (5)
C9—C10	1.356 (6)	C19—H19A	0.96
C9—H9	0.93	C19—H19B	0.96
C10—C11	1.389 (6)	C19—H19C	0.96
C10—H10	0.93	Cl1—Sn1	2.3478 (16)
C11—C15	1.520 (6)	Cl2—Sn1	2.4644 (17)
C12—C14	1.497 (7)	N1—H1	0.833 (18)
C12—C13	1.505 (7)	O1—Sn1	2.375 (3)
C12—H12	0.98		
O1—C1—C2	121.9 (4)	C12—C14—H14A	109.5
O1—C1—C4	118.8 (4)	C12—C14—H14B	109.5
C2—C1—C4	119.3 (4)	H14A—C14—H14B	109.5
C3—C2—C1	125.3 (4)	C12—C14—H14C	109.5
C3—C2—H2	117.3	H14A—C14—H14C	109.5

C1—C2—H2	117.3	H14B—C14—H14C	109.5
N1—C3—C2	122.7 (4)	C17—C15—C11	112.4 (5)
N1—C3—C5	117.4 (4)	C17—C15—C16	110.5 (5)
C2—C3—C5	119.9 (4)	C11—C15—C16	109.5 (4)
C1—C4—H4A	109.5	C17—C15—H15	108.1
C1—C4—H4B	109.5	C11—C15—H15	108.1
H4A—C4—H4B	109.5	C16—C15—H15	108.1
C1—C4—H4C	109.5	C15—C16—H16A	109.5
H4A—C4—H4C	109.5	C15—C16—H16B	109.5
H4B—C4—H4C	109.5	H16A—C16—H16B	109.5
C3—C5—H5A	109.5	C15—C16—H16C	109.5
C3—C5—H5B	109.5	H16A—C16—H16C	109.5
H5A—C5—H5B	109.5	H16B—C16—H16C	109.5
C3—C5—H5C	109.5	C15—C17—H17A	109.5
H5A—C5—H5C	109.5	C15—C17—H17B	109.5
H5B—C5—H5C	109.5	H17A—C17—H17B	109.5
C11—C6—C7	121.8 (4)	C15—C17—H17C	109.5
C11—C6—N1	118.9 (4)	H17A—C17—H17C	109.5
C7—C6—N1	119.2 (4)	H17B—C17—H17C	109.5
C8—C7—C6	117.9 (4)	Sn1—C18—H18A	109.5
C8—C7—C12	120.8 (4)	Sn1—C18—H18B	109.5
C6—C7—C12	121.3 (4)	H18A—C18—H18B	109.5
C9—C8—C7	121.0 (4)	Sn1—C18—H18C	109.5
C9—C8—H8	119.5	H18A—C18—H18C	109.5
C7—C8—H8	119.5	H18B—C18—H18C	109.5
C10—C9—C8	120.6 (4)	Sn1—C19—H19A	109.5
C10—C9—H9	119.7	Sn1—C19—H19B	109.5
C8—C9—H9	119.7	H19A—C19—H19B	109.5
C9—C10—C11	120.8 (4)	Sn1—C19—H19C	109.5
C9—C10—H10	119.6	H19A—C19—H19C	109.5
C11—C10—H10	119.6	H19B—C19—H19C	109.5
C6—C11—C10	117.9 (4)	C3—N1—C6	125.1 (3)
C6—C11—C15	122.0 (4)	C3—N1—H1	117 (3)
C10—C11—C15	120.1 (4)	C6—N1—H1	117 (3)
C14—C12—C13	111.3 (5)	C1—O1—Sn1	136.4 (3)
C14—C12—C7	109.7 (4)	C18—Sn1—C19	142.9 (3)
C13—C12—C7	113.5 (4)	C18—Sn1—Cl1	107.61 (16)
C14—C12—H12	107.3	C19—Sn1—Cl1	107.20 (18)
C13—C12—H12	107.3	C18—Sn1—O1	88.67 (17)
C7—C12—H12	107.3	C19—Sn1—O1	84.0 (2)
C12—C13—H13A	109.5	Cl1—Sn1—O1	81.74 (9)
C12—C13—H13B	109.5	C18—Sn1—Cl2	94.04 (15)
H13A—C13—H13B	109.5	C19—Sn1—Cl2	94.78 (18)
C12—C13—H13C	109.5	Cl1—Sn1—Cl2	95.82 (6)
H13A—C13—H13C	109.5	O1—Sn1—Cl2	176.81 (8)
H13B—C13—H13C	109.5		
O1—C1—C2—C3	−1.0 (7)	C8—C7—C12—C14	87.0 (6)
C4—C1—C2—C3	178.8 (5)	C6—C7—C12—C14	−91.4 (5)
C1—C2—C3—N1	0.4 (7)	C8—C7—C12—C13	−38.2 (6)

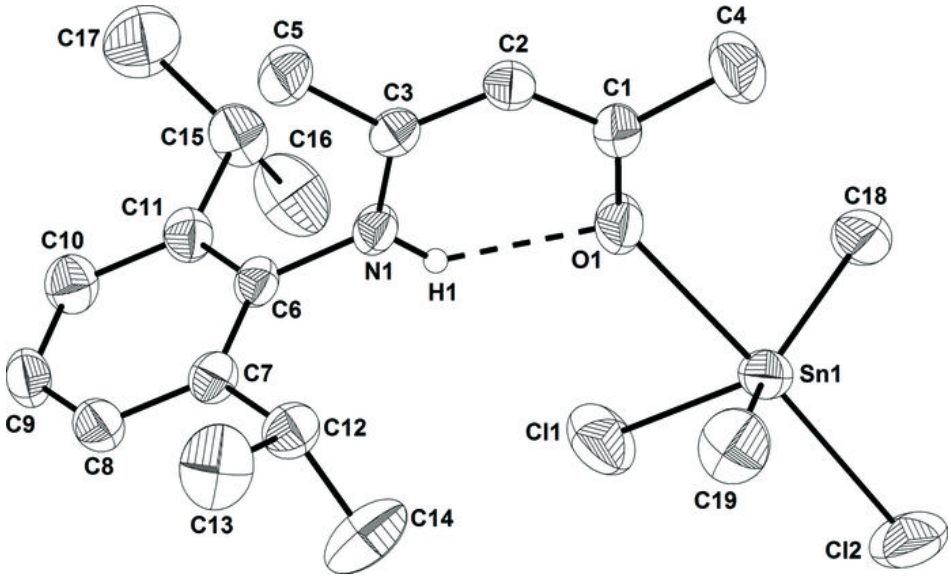
C1—C2—C3—C5	179.4 (4)	C6—C7—C12—C13	143.3 (5)
C11—C6—C7—C8	0.0 (6)	C6—C11—C15—C17	−140.8 (5)
N1—C6—C7—C8	179.3 (4)	C10—C11—C15—C17	41.3 (7)
C11—C6—C7—C12	178.5 (4)	C6—C11—C15—C16	96.0 (5)
N1—C6—C7—C12	−2.2 (6)	C10—C11—C15—C16	−81.8 (6)
C6—C7—C8—C9	−1.3 (6)	C2—C3—N1—C6	−179.2 (4)
C12—C7—C8—C9	−179.8 (4)	C5—C3—N1—C6	1.7 (6)
C7—C8—C9—C10	1.0 (7)	C11—C6—N1—C3	86.6 (5)
C8—C9—C10—C11	0.6 (7)	C7—C6—N1—C3	−92.8 (5)
C7—C6—C11—C10	1.6 (6)	C2—C1—O1—Sn1	−156.7 (3)
N1—C6—C11—C10	−177.7 (4)	C4—C1—O1—Sn1	23.6 (7)
C7—C6—C11—C15	−176.3 (4)	C1—O1—Sn1—C18	−68.9 (5)
N1—C6—C11—C15	4.4 (6)	C1—O1—Sn1—C19	74.7 (5)
C9—C10—C11—C6	−1.8 (7)	C1—O1—Sn1—Cl1	−176.9 (5)
C9—C10—C11—C15	176.1 (4)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1···O1	0.83 (3)	2.03 (4)	2.662 (5)	133 (4)
C8^{i}—H8^{i}···Cl1	0.93	2.91	3.695 (4)	143
C17^{ii}—H17B^{ii}···Cl2	0.96	2.89	3.783 (6)	155
C19^{iii}—H19C^{iii}···Cl2	0.96	2.94	3.700 (6)	137

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*+2; (ii) *x*+1, *y*−1, *z*; (iii) −*x*+2, −*y*+1, −*z*+1.

Fig. 1


C17
C5
C2
C4
C3
C1
C15
C16
C18
C11
C10
N1
O1
H1
C6
C7
Sn1
C9
C12
Cl1
C8
C19
C13
C14
Cl2